Lesson 1: “Better Living Through Chemistry?”: Exploring the Role of Plastic in Post-WWII American Culture

Understanding Goals:

In this lesson, students will:

• Learn why plastics became such a pervasive part of American society during the postwar era.
• Explore early concerns about the potential effects of plastics on workers’ health
• Examine why different individuals and organizations evaluated and perceived risk differently

Lesson Overview:

The lesson will begin by first assessing what students know about synthetic chemicals associated with plastic. Recently, there has been growing concern about how certain chemicals associated with plastic, including PVC, bisphenol-A, and phthalates affect human and environmental health. Some students may have heard about these concerns; others may know very little about the subject. By first assessing what students know about the topic, the teacher will gain a deeper understanding of the knowledge that students bring to the classroom. It will provide a starting point to help students think about how different understandings of plastic’s effects on health are generated and reinforced. These preliminary conversations will demonstrate the point that how we know what we know is based on a wide variety of sources. It will help students recognize that we do not always evaluate these sources critically or thoroughly.

After assessing students’ initial ideas though small and large-group discussion, the teacher will provide some background on the history of the boom in the chemical industry and the production of plastics during the postwar era. Background information and additional resources are listed below.

Students will also have an opportunity to analyze primary documents from the 1950s and ‘60s to reinforce the idea that plastics became a significant part of the American society as a result of several of economic, cultural, and political factors. Then students will analyze primary documents to explore how and why new understandings about the relationship between and workers’ health emerged. This introductory lesson will help students understand the origins of debates over the regulation of synthetic chemicals.

Background Information:

From the cars we drive, to the computers we work on, to the water we drink, plastic has become a pervasive part of our society today. It has become such a ubiquitous part of our lives that is difficult to imagine a world without plastic. This introductory
Plastic was first invented in the 1860s by Alexander Parkes in London. Known as Parkesine, the first plastic was made from cellulose and other organic materials. It served the same functions as rubber, but it was cheaper to produce. In 1869, an American chemist named John Wesley Hyatt produced the first thermoplastic—a substance shaped under pressure and heat that retains its shape when cooled. Known as celluloid, it was first used to replace ivory in billiard balls and provided materials for the first moving pictures. In 1907, Leo Baekeland developed Bakelite—a form of thermoplastic that was even more durable than previous plastics. In addition to a variety of household uses, the U.S. military used bakelite to produce lightweight weapons. War machinery improved with bakelite was extensively used in WWII. Also, prior to WWII, other notable thermoplastics such as polyvinyl chloride (PVC), polystyrene, and polyethylene came into production. As demand for resources increased during the war, these new plastics began to play an increasingly important role in American society.

To support the war effort, engineers created a wide array of new products and lightweight materials—including everything from weapons to food packaging (Saran Wrap was a WWII invention). After the war, the production of plastic products for civilian uses increased exponentially. Plastics became a versatile symbol of the benefits of the growing industry. “Better living through chemistry” became Du Pont’s slogan and other chemical companies such as Dow, Monsanto, and Union Carbide and others also launched massive advertising campaigns. These campaigns depicted plastic as a savvy substance that could be molded to meet the many needs of modern society.

At the same time that chemical companies claimed that new plastic goods would provide solutions to an array of modern problems, the industry was also becoming aware of some of the deleterious effects that certain had on human health. We will explore the emergence of these new understandings in the latter part of this lesson.

Preparation:

This lesson involves small group activities and large group discussion. The teacher will provide some history on plastic between activities and should be familiar with the material in the background information section. There are also additional resources listed at the end of the lesson where teachers can go for more information on the subject. Documents used for primary analysis can be downloaded directly off this site. Teachers can click on the links and print out copies of the documents for students to use in class.

Activity 1.1: Assessing Students’ Understandings about Plastic: From initial uses to potential effects
Assess students’ existing knowledge about plastic’s origins and its effects on human and environmental health. Ask students to reflect silently and then write down their ideas on the following questions:

1. Why do you think plastic was initially created? What purpose did it serve? Where was it first invented and used? By whom?
2. Why do you think people would have welcomed the invention of plastic in the early twentieth century? In the mid-twentieth century?
3. What do you know about how plastic affects human health? How does plastic affect the environment?

Assure students that you don’t expect them to have the “right” answer at this point. Emphasize that you want them to think about plausible initial uses and past problems that plastics may have been invented to solve. Also, tell them that the point of this introductory activity is to assess the range of ideas that are out there about the relationship between plastic and health.

Large group discussion:

Students can all share their ideas by going around the whole class in a circle, or the teacher can ask a few willing individuals to share. Students will most likely come to class with a range of ideas based on an equally wide range of sources. When discussing their ideas about how plastic affects health, ask them where they got their information. This large-group discussion need not be too long or involved; simply emphasize the wide range of initial ideas and sources of where we get our information.

Activity 1.2: Exploring Multiple Causes of the Postwar Boom in the Plastic Industry

The teacher will explain that the postwar growth of the chemical industry was the result of a variety of economic, cultural, and political factors (i.e. the boom in petroleum production, the portrayal of plastic in popular media, the faith in science and technology, and the power and influence of the chemical companies). Depending on the amount of class time available and familiarity of the teacher with the subject, this talk can last from 8-15 minutes. In addition to the background material provided, teachers can also access other resources at links provided at the end of this unit.

Analyzing Primary Sources:

After the teacher explains some of this background material, students will have an opportunity to analyze a range of documents from the 1950s and ‘60s about the status of plastic in American society. Have students count off by two and explain that the “ones” will analyze document A, the “twos” will analyze document B, “threes” analyze C. Each group will analyze one of the three documents. Document “B” involves two separate advertisements, since the other documents will take a little
longer to read and analyze. Depending on class size, there may be two or three groups assigned to each document. Electronic versions of these documents are available at the links provided below.

**Documents for Activity 1.2:**


B. Two advertisements from chemical companies:
   - “This is the World of Union Carbide,” Display ad in the *New York Times*, January 9, 1961.


**Small group discussion:**

Have the students discuss the following questions:

1. Who was/were the authors of this document?
2. Who was the intended audience of this document?
3. Why do you think this document was written? What evidence does the document contain that makes you think it was produced for that reason? Please refer to specific quotes or details from the document.
4. Based on this document, why do you think plastic came to hold such an important place in postwar American culture?
5. What does this document reveal about the status of plastic and plastic production in contemporary American society?
6. What does it NOT say? That is, what does this document hide—intentionally or unintentionally? List three questions that are left unanswered by the document.

After talking about the perspectives offered by the different documents in small groups, the class will reconvene in a large group. A student representative from the small group will explain how their document addresses the question: “Why did plastics become such an important part of American society after WWII?” Ask students to think about how different economic, cultural, and political factors influenced the rise of the plastics industry.

**Activity 1.3: Analyzing Understandings of Plastics’ affect on Workers’ Health**

The purpose of this activity is get students to think about ways that the postwar boom in plastic production affected human and environmental health. It will explore understandings that developed in the 1950s about the relationship between plastics and health.
In their same small groups, students will examine Documents A, B, and C. Because documents A and B will take longer to read and analyze, document C contains two different sources for students to work on. These documents illustrate different understandings of how the production of plastic affected (or did not affect) human health. Depending on time available, this activity can be done in class or assigned to students to complete at home.

**Documents for Activity 1.3:**

A) D. Kenwin Harris, “Health Problems in the Manufacture and Use of Plastics,” *British Journal of Industrial Medicine*, vol. 10, 1953. Students should read pages 255 to the top of page 261. Also read the “Discussion” and “Summary” sections on pages 266 and 267. Skim intervening section and note images of affected workers.


C) Union Carbide Corporation, Inter-Company Correspondence, November 24, 1959.

Ask students to reflect and write their responses to the following questions:

1. Who were the authors of this document? What do you know—or think you know—about the authors?
2. Who was the intended audience of this document?
3. Why do you think this document was written? What evidence does the document contain that makes you think it was produced for that reason? Please refer to specific quotes from the document.
4. What does the document say about the potential affects of plastics and chemicals involved in plastic production on human health?
5. List three questions that are left unanswered by the document.

**Large-group Discussion:**

After working independently with their assigned document, students should share their analyses of the documents. After each document has been discussed ask students why the author of the 1959 Union Carbide memo did not want the memo quoted? What is the significance of the memos and what are some of the possible outcomes of it? In this discussion, students should make connections back to the previous lesson and talk about the cultural, economic, and political factors that made plastics so popular in the postwar era. They should also begin to think about the issue of how to deal with scientific uncertainty in the face of regulating potentially harmful chemicals. If Activity 1.3 is assigned as homework, this large-group discussion can be used as a starting point for the following lesson.
Activity 1.4: Homework

Students should read portions of Rachel Carson’s *Silent Spring*, answer the following questions, and be prepared to discuss their ideas in lesson two.


**Homework Questions:**

1. What is the purpose of the book?
2. Why was Carson so concerned about ways in which humans have changed the environment since WWII?
3. Carson argues that Americans have a right not to be exposed to toxic chemicals. Do you agree or disagree? Why and under what conditions?
4. In your mind, whose duty is it to protect people from toxic chemicals? Why?
5. How do you think the chemical industry received the arguments Carson makes in *Silent Spring*?
6. Is *Silent Spring* a primary sources or a secondary source? Why do you think so? Explain your answer.

Lesson 1 Reinforcement Activity: Analyzing a WWII-era Film by the Chemical Industry

If there is time available, before exploring the documents and questions in Activity 1.2, students can view a film created by the chemical industry in the 1940s to inspire young scientists to become chemists. It demonstrates how chemical companies used media to portray a wide variety of ways in which chemicals will benefit postwar society. After viewing the 9-minute video, ask students to respond to the following reflection questions:

1. What is the central message(s) of this video?
2. How do you think the filmmakers wanted the audience to respond?
3. What does this type of media say about the topic that would not be conveyed by a written source? Please be specific.
4. Write two questions that are left unanswered by the film.

**Document:**


**Additional Teacher Resources:**

The American Chemistry Council’s History of Plastic:
Lesson 2: Exploring the Hidden Costs of “Better Living”

Understanding Goals:

In this lesson, students will:

- Learn how debates over the effects of plastics and new synthetic chemicals on workers’ health were connected to the growing consumer, public health, and environmental movements.
- Explore how the creation of new chemicals led to new knowledge and debates about how to regulate potentially harmful chemicals in the face of scientific uncertainty.

Lesson Overview:

This lesson investigates the rise of the synthetic chemical industry and explores debates over how new chemicals associated with the production of plastic affected human health. Activities help students make connections between the emergence of the debates over plastics and the broader environmental, public and consumer health movements. During the 1950s, as concern about the potentially harmful effects of new synthetic chemicals arose, the federal government began to consider measures for regulation. In the beginning portion of this lesson, students will explore testimony from the Chemicals in Food Products Hearings of 1950-51 and the Delaney Clause of 1958. Then they will lean about Rachel Carson’s *Silent Spring* and the backlash that the work received from the chemical industry. This lesson will help students gain a deeper understanding of historical debates about synthetic chemicals associated with the production of plastic.

Background Information:

This lesson demonstrates that concerns about the effects of chemicals on people and the environment preceded *Silent Spring*, but it also emphasizes that Carson’s work was extremely important for communicating scientific knowledge to a widespread audience. Students will explore debates from the 1950-51 Chemicals and Foods Hearings, learn about the Delaney Clause, and analyze *Silent Spring* and criticisms of Carson’s work.

*Background on Chemicals and Foods Hearings of 1950-51*

During the 1950s, a young representative from Queens, NY named James Delaney led a series of hearings concerning the potential harm of chemical additives in food and related food products, such as plastic wraps and containers. Testimony at the hearings came from a variety of individuals and organizations that were becoming concerned about the potential health effects of plastics on health. In this activity, students will analyze excerpts offered by different perspectives during the hearing. Working in the same small groups, students will examine the different ideas about the
role of chemicals in American life and growing concerns about how they might affect health.

Background on the Delaney Clause:

The federal dialogue that began with the 1950-51 Chemicals and Foods Hearing eventually led to an amendment of the Federal Food, Drug and Cosmetic Act in 1958. The amendment created new regulatory policies for “food additives” and included the controversial Delaney Clause. The Delaney Clause prevented the Food and Drug Administration from approving the use of chemical additives to food that had been proven to cause cancer in humans or other animals. This included packaging material and food related-materials like can liners and plastic wrap. According to the 1958 amendment, the food products and packaging industry had to prove to the FDA that all food additives were safe. At that time, there were only a few chemicals in use that had been proven to cause cancer in animal studies, so sponsors of the measure did not think it would have a major impact.

Background on Silent Spring and its Critics:

Rachel Carson’s *Silent Spring* played an especially important role in bringing widespread attention to some of the negative effects of chemicals on human and environmental health. When it was published in 1962, it became an instant bestseller. Carson clearly articulated a growing fear among scientists and activists that synthetic chemicals could have hazardous effects on human health and environmental systems. Because of this public awareness, chemical companies were increasingly put on the defensive. As activists, scientists, and the federal government became increasingly concerned about the potential long-term harm that these new chemicals might inflict on people and the environment, the chemical industry enacted a massive public relations campaign to address these concerns and to undermine arguments made by chemical critics. For more background on *Silent Spring*, Rachel Carson, and the debates that her work inspired, see the links listed in the additional resources section of this lesson.

Activity 2.1: Examining the Chemicals in Foods Debates of 1950-51

Depending on the degree of students’ familiarity with the topic, the teacher should talk a bit about the rise of public concern about occupational and environmental health. Links to background information on the topic are listed at the end of this lesson plan. The teacher should explain that the same chemical companies who were producing new plastic food wraps and containers were also producing other new consumer goods, such as pesticides, fertilizers, and fungicides. Consumer health and environmental activists became increasingly concerned about the effects that these new chemicals—used for both agricultural and non-agricultural purposes—would have on human and ecological health.
Have the class read Delaney’s 1951 report on the use of chemicals in food products. Then have them read the meeting minutes of the Plastics Committee of the Manufacturing Chemists Association, Inc. regarding the Chemicals in Foods hearings. Ask the class the following questions:

1. Why does Delaney think that the investigation into the use of chemicals in food products is necessary? What is the problem? What evidence does he use to outline this problem?
2. Who does Delaney think should be responsible for regulating the use of chemicals in food products? Why does he think that?
3. Who does the Plastics Committee of the Manufacturing Chemists Association think should be responsible for regulation? Why do they think that?
4. If you were alive in the 1950s, who do you think you would agree with? Why?

If there is time, the class can hold a debate simulating the 1950-51 Chemicals in Foods hearings. The teacher can split the class in half and have each group prepare an argument for who should be in charge of regulating chemicals in food products (government or industry). Students should base their arguments on documents in this lesson as well as from lesson one. In the debate, each group will have two minutes to present their argument, including evidence to support their argument based on the testimony they read. They should not read actual testimony verbatim, but rather, summarize the essence of the argument and articulate it in their own words. As each group presents, members of other group should take notes. These notes will be used in the open session. After each group has made their opening statement, acting as moderator, the teacher will open the floor to questions, comments, and rebuttals. Encourage the students to make connections back to themes and ideas from lesson 1 (i.e. the cultural, economic, and political significance of plastic in postwar American culture). The teacher should make sure that each group gets equal time to pose and respond to questions.

After the open session, the teacher should summarize by stating that in accordance with the historical record, the 1951 Chemicals in Foods investigation came to a close without definitive regulatory action for the future. The teacher should note that Delaney’s report raised important questions and concerns and the federal government would be taking the matter into further consideration. Furthermore, the re-enactment helps demonstrate that concerns about chemicals and the environment began before the official 1970s environmental legislation and the widespread public outcry caused by Rachel Carson’s *Silent Spring*.

**Documents for Activity 2.1:**


Activity 2.2: Understanding and Interpreting the Delaney Clause

As a class, have students read the 1958 Delaney Clause out loud.

"No additive shall be deemed to be safe if it is found to induce cancer when ingested by man or animal, or if it is found, after tests which are appropriate for the evaluation of the safety of food additives, to induce cancer in man or animal."

- The Delaney Clause, from the 1958 Food Additives Amendment (Section 409) to the Federal Food, Drug, and Cosmetic Act

Large Group Discussion:

Ask students to explain, in their own words, what they think the provision means. What are some of its strengths? What are some of its weaknesses? What measures would be required to implement the amendment? Can they think of any examples of regulation that might have been the result of the Clause or related a related principle?

Although there were several loopholes in the Delaney Clause (see Richard Merrill, 1997 and the Congressional Research Service’s report by clicking on the links provided below), its passage indicated that the consumer health movement had begun to gain traction. People were becoming concerned about the abundance of synthetic chemicals in their food, water, and everyday products. It is important to emphasize to students that although many people think Silent Spring was the book that started the environmental movement by catalyzing public awareness about the effects of chemicals on human and ecological health, prior to its publication, several scientists and activists had also been concerned about the issue.

Activity 2.3: Putting Concerns about Plastics in Historic Context: Silent Spring and Carson’s Critics

Students should come to class having read the first three chapters of Silent Spring. The teacher should explain a bit about the background and significance of the book. As a class, review and discuss students’ responses to the homework questions about Silent Spring:

7. What is the purpose of the book?
8. Why was Carson so concerned about ways in which humans have changed the environment since WWII?
9. Carson argues that Americans have a right not to be exposed to toxic chemicals. Do you agree or disagree? Why and under what conditions?
10. In your mind, whose duty is it to protect people from toxic chemicals? Why?
11. How do you think the chemical industry received the arguments Carson makes in *Silent Spring*?
12. Is *Silent Spring* a primary source or a secondary source? Why do you think so? Explain your answer.

After students discuss the book and explore some of their own reflections, the teacher should present them with some of the criticism of the book. Pass out copies of William Darby’s review of *Silent Spring* from *Chemical & Engineering News* (1962) and have students read the review.

**Document for Activity 2.3:**


In small groups, have students respond to the following questions:

1. What is Darby’s main criticism of *Silent Spring*?
2. How does he think Carson “confuse the information” and bias the book with her opinions (p2)?
3. Darby makes a distinction between “occupational and residue hazards”—the former are isolated effects of chemicals on workers in manufacturing plants, the latter refers to residues of chemicals on foods and the environment. He claims that while there had been some trouble with health hazards at the workplace, in the broader environment there had been no examples of “injury resulting to man from these residues (p3).” What do you think about this claim?
4. How do Carson’s and Darby’s ideas about the relationships between human and the environment differ? How are they similar?

**Activity 2.4: Homework Assignment:**

Read *Deceit and Denial*, Chapter 5-7, pp. 139-234.

The following quotations come from this reading. Think carefully about their meaning and context and respond to the following questions. Responses for both questions need not exceed three pages.

1. On page 210, Markowitz and Rosner argue that “The vinyl chloride crisis substantially blurred the line between occupational and environmental dangers.” What does this statement mean? What evidence do the authors use to justify this argument? What are its implications?

2. On page 216, Andrea Hricko, a staff assistant working for Wolfe Sidney’s Health Research Group, was quoted as saying, “Unfortunately, [epidemiological proof] is always retrospective in nature and can only be accumulated after the harmful effects
have already manifested themselves.” She believed that all new chemicals needed to be tested before they were widely produced and distributed. Do you agree with this argument? Why or why not? What would be the benefits (and to whom) of thoroughly testing all new synthetic chemicals before their production and distribution? What would be the drawbacks of such a policy?

**Reinforcement Activity: Exploring the Plastics Industry’s Response to Environmental Concern**

In the early 1970s, the plastic industry faced growing criticism from scientists, activists, and increasingly, federal legislators. One representatives of the industry complained that he hated to go to cocktail parties because he always seemed to get into a “big environmental argument with somebody’s wife (See the transcript of Harding’s 1971 talk, p9).” In order to ameliorate concerns about plastics’ effect on human and environmental health, they launched a “grass roots” program. Plastic advocated a “persuade your neighbor” campaign where individuals would spread positive information about the plastic industry as being “part of the solution—not pollution.”

In this activity, half of the class will analyze documents produced by the plastics industry that aimed to address the public’s growing concerns about environmental pollution. Students will analyze two different documents and answer the following questions about each:

1. Who is the author of this document? Who is the intended audience?
2. What are the main problems that each author addresses? How does the author talk about these problems?
3. What actions do the authors advocate for as means to ease public concerns about plastic? Be specific.
4. Do you think these actions would be effective? Why or why not? What are the strengths of these actions? What are their weaknesses?

**Documents for Reinforcement Activity:**


**Additional Resources:**


*On Rachel Carson and Silent Spring:*
http://www.rachelcarson.org/
[http://www.chatham.edu/RCI/](http://www.chatham.edu/RCI/)
Lesson Three: The PVC Story: Exploring Responses, Regulation, and Research Validity

Understanding Goals:

In this lesson, students will:

- Learn how different interests develop knowledge and approach decision-making in regards to toxic chemicals
- Consider different historical perspectives on what constitutes ‘sound science’ and its role in regulating potentially toxic chemicals

Lesson Overview:

This lesson helps students understand how knowledge about toxic chemicals was created, used, and framed in different arguments for and against the regulation of polyvinyl chloride (PVC). Students will examine how competing claims about the possible risks to health played out in debates about the regulation of PVC. By exploring different perspectives, students will gain a better understanding of why different interests—activists, scientists, industry representatives, workers—developed different interpretations of the same phenomenon.

This lesson will also provide opportunities for students to understand how historians evaluate the reliability and effectiveness of primary sources. Activities will also help students understand how historical knowledge is used in contemporary debates.

Background Information:

Controversy over the health effects of PVC grew during the 1970s as the deaths of workers in different chemical plants were linked to the carcinogenic properties of a PVC monomer. During the beginning stages of the controversy, because there was so little known about the effects of new synthetic chemicals, it was difficult to make claims that PVC caused cancer. Scientists, workers, industry representatives, and the press attributed causality and approached regulation in different ways. These different understandings lay at the heart of debates over how to regulate PVC.

In 2002, historians Gerald Markowitz and David Rosner published *Deceit and Denial: The Deadly Politics of Industrial Pollution*. Two chapters of the book explain how the chemical industry tried to hide and distort studies that proved that the vinyl chloride monomer was carcinogenic. The authors relied on evidence gathered as a result of legal proceedings against the industry. A lawyer from Lake Charles, Louisiana named Billy Baggett, Jr. had collected a warehouse full of internal memos, meeting minutes, and correspondence produced by the Manufacturing Chemists Association. These documents not only helped produce *Deceit and Denial*, but they were also used in a 2001 Bill Moyers’ Special Report called *Trade Secrets* and a 2002 documentary by Judith Helfand and Dan Gold called *Blue Vinyl*. Many of the
documents can be accessed at the Chemical Industry Archives at the link listed in the Additional Resources section of this lesson.

In response to the media attention received by these productions, the chemical industry has criticized all three for misrepresenting evidence and distorting the truth. They claim that these liberal media sources are sensationalizing the issue and playing on people’s preconceived notions of the chemical industry. After analyzing some of the same primary documents used in these productions, student will have a chance to draw their own conclusions. Then they will compare their ideas with the historians, the filmmakers, and their critics.

**Activity 3.1: Analyzing Causes for Concern**

Using primary sources, students will analyze a range of documents to gain a deeper understanding of how different perspectives—scientists, chemical industry executives, and chemical workers and their families—understood the effects of chemicals. Students will break up into small groups of three people. Each group will get a copy of all three documents listed below. Each student will analyze a different document and answer the questions listed below. When all group members finish their analyses, they will report to their small group.

**Documents for Activity 3.1:**


**Questions:**

1. Who was the author(s) of this document? What do you know about the author(s)?
2. Who was the intended audience of this document?
3. Why do you think this document was created? What evidence does the document contain that makes you think it was produced for that reason? Please refer to specific quotes from the document.
4. What does the document say about the potential affects PVC on human health?
5. List three questions that are left unanswered by the document.

After each student reports to the small group, students should compare the different perspectives. Ask the small groups to consider how each author would address questions about regulating potentially harmful chemicals. Have students explain to
each other what kinds of measures that they think P.L. Viola, the Italian scientist; R.L. Wheeler, jr., the Union Carbide executive; and Valerie Arthur the deceased worker’s 15 year-old-daughter would advocate for in dealing with the PVC issue. Why do students think the different perspectives would be likely to take those actions? How do their decisions relate to the principles behind the Delaney Clause?

Activity 3.2: Comparing Interpretations

After the small group discussion, reconvene as a class. Have the students reflect on their own interpretations of the PVC controversy and compare it what they read in Deceit and Denial. Specifically, on page 178, Gerald Markowitz and David Rosner write that the chemical industry successfully deceived its workers and the public by “hiding its information about cancer from the government” and “deflecting national attention away from the potential hazards of thousands of mostly untested new chemicals and of vinyl chloride in particular.” Do you agree with the historians’ interpretation? Why or why not?

Depending on time and availability, the teacher may want to show students excerpts from Bill Moyers’ Trade Secrets and/or Judith Helfand’s Blue Vinyl. Moyers and Helfand come to similar conclusions as Markowitz and Rosner; namely, that the chemical companies withheld information about the risks of PVC to workers, surrounding communities, and the general public. Have students compare their own understandings to these additional sources. Do they agree with the central claims of these documentaries? What questions do they have for the filmmakers? What are the strengths of these documentaries? What are their weaknesses?

Activity 3.3: Knowledge on Trial: Evaluating Criticism and Debating Research Validity

Like the Trade Secrets and Blue Vinyl documentaries, the publication of Deceit and Denial generated controversy and has had important legal impacts. In 2005, Markowitz was to serve as an expert witness in a trial in Mississippi where a chemical worker was planning to sue his former employer for knowingly exposing him to the polyvinyl chloride monomer that gave him cancer. In response to Markowitz’s subpoena, the chemical industry hired lawyers and a professor from Rutgers University named Philip Scranton to discredit his research.

Have students read Jon Weiner’s article in The Nation that appeared on February 7, 2005. Then divide the class in half. Give one group copies of Scranton’s criticism of Deceit and Denial; give the other half Markowitz and Rosner’s rebuttal. These reports are both fairly long so instead of having students thoroughly read them cover-to-cover, they should browse the main points of each memo. After students have a chance to read and discuss their documents, they should be prepared to debate the other half of the class. They can draw from the text of the documents provided below as well as other sources.
The debate will focus on the following questions: Is Markowitz and Rosner’s research valid? Should Markowitz be allowed to provide testimony in the lawsuit against the chemical company?

Documents for Activity 3.3:

C. David Rosner and Gerald Markowitz, “Response to Philip Scranton’s Report on Deceit and Denial: The Deadly Politics of Industrial Pollution.”

Activity 3.3: Homework Assignment:

Finish reading Deceit and Denial, Chapters 8-Conclusion, 234-306. After reading the book and hearing some of the criticisms of the work presented in the class debate, students should write a 2-3 page response paper. The paper should contain three parts: 1) an explanation of Markowitz and Rosner’s main points; 2) an explanation of Scranton’s criticism; and 3) an explanation of which perspective they thought was the most persuasive and why.

Reinforcement Activity 1: Understanding Epidemiological Studies

In the late 1980s, a pharmacist in the St. Gabriel neighborhood named Kay Gaudet began to notice that a number of her customers were receiving medications having to do with various reproductive problems. She also sensed that the number of miscarriages in her neighborhood was unusually high. St. Gabriel was one small community in “cancer alley,” an 85-mile corridor along the Mississippi River between Baton Rouge and New Orleans that included 136 petrochemical plants and 7 oil refineries. Gaudet suspected that the reproductive issues she had noticed were connected to the pollution from the chemical companies. She decided to conduct an informal study to see how the rate of miscarriages in the St. Gabriel community compared with other areas. Gaudet found that of the 2,100 people who lived in St. Gabriel, there were 75 miscarriages between 1985 and 1988.

Although Gaudet’s findings reinforced her belief that the chemical plants were responsible for the threats to her community’s health, a study published in 1989 by the Tulane Department of Public Health and Tropical Medicine found that the rate of miscarriage in St. Gabriel and two surrounding communities—Carville and Sunshine—was statistically not higher than the state average.

The discrepancy between the two studies provides an opportunity for students to consider some of the challenges and limitations of epidemiological studies. In this activity, students will read about each study and learn to think critically about these kinds of studies and their role in regulating potentially hazardous chemicals.
Documents for Reinforcement Activity 1:

B. LuAnn E. White et al., “St. Gabriel Miscarriage Investigation, East Bank of Iberville Parrish, Louisiana,” Final report of study conducted by Tulane University, School of Public Health and Tropical Medicine, New Orleans, September 27, 1989.
C. Deceit and Denial, p.256-257 for Gaudet’s criticism of the Tulane Study.

Discussion Questions:

1. How do the Gaudet and Tulane studies differ? How are they similar?
2. Why do the studies come to different conclusions?
3. What are some of the methodological limitations of Gaudet’s study?
4. What are some of the methodological limitations of the Tulane study?
5. What are some of Gaudet’s criticisms of the 1989 Tulane study?
6. Do you think her criticisms are valid? Why or why not?
7. If you were the Governor of Louisiana, what kinds of action would you take to deal with the problems associated with “cancer alley”? Which study would you use to justify your actions? Why?

Reinforcement Activity 2: Mapping Cancer Alley

In this activity, students will examine the locations of petrochemical plants in cancer alley and explore demographic data about surrounding communities. Who are the people who live near our country’s largest concentration of petrochemical industries? Why do they live there? The purpose of this spatial analysis is to encourage students to think how questions of social and environmental justice play into debates over the regulation of synthetic chemicals.

Procedure: Click on the U.S. National Library of Medicine’s Environmental Health E-Maps site (TOXMAP): http://toxmap.nlm.nih.gov/toxmap/home/welcome.do. In the search fields, select “Louisiana” as the state. A map showing the locations of all facilities that are included in the Toxic Release Inventory will appear. Students will be able to visually see the large concentration of toxic plants along “cancer alley.” Have students zoom in on this part of the state by clicking on the “Zoom In +” icon. To the right of the map there is a section called “Apply to this map.” This gives users options to add about race, income, and health of the area under investigation.

1. Have student clink on “U.S. Census Data” and in the list of data from “2000 Race,” have them select a category to investigate. Then have them clink on “Submit.” The map will then show this demographic data with the location of the TRI facilities superimposed on it. Ask students to examine the relationships between concentration of facilities and this demographic data. Do they see any patterns?
2. Then have them return to the right side of the map and click on “Income Data.” Have them click on the Per Capita Personal Income value for a given year. Again, ask students to examine the relationships between concentration of facilities and this demographic data. Do they see any patterns?

3. Repeat this with the “Health Data” section and have students explore different categories under the “Mortality, Cancer 2000-2004” field. Again, ask students to examine the relationships between concentration of facilities and this demographic data. Do they see any patterns?

Ask students to reflect on what these maps. Where does the information used to generate these maps come from? What general patterns do they seem to show? How would you go about testing relationships between the location of plants that release toxins and surrounding communities? What are the strengths of mapping tools such as TOXMAP? What are their weaknesses? What kinds of questions do these spatial analyses raise?

Have them repeat steps 1-3 for their home communities. How do their home communities compare to “cancer alley”? Do they notice any similar patterns? What are the major differences?

**Additional Resources:**


Lesson 4: Exploring the Future History of Plastics: Contemporary Debates about Bisphenol-A and Phthalates

Understanding Goals:

In this lesson, students will:
- Analyze current controversies over the regulation of bisphenol-A.
- Make connections between current issues and debates and themes discussed in previous lessons.

Lesson Overview:

In the first three lessons, students learned how and why synthetic chemicals became such a pervasive part of American life. They considered how different perspectives developed different understandings of the effects of synthetic chemicals on human and environmental health. Activities also helped students understand the complexity of regulating synthetic chemicals. In this final lesson, students will grapple with questions about risk and uncertainty by engaging in a case study about the regulation of bisphenol-A and phthalates.

Background:

Since the 1950s, bisphenol-A and a number of phthalates have become ubiquitous in the environment and in our bodies. Though it was invented in the nineteenth century, the mass-production of bisphenol-A began after WWII as a component of polycarbonate plastic—an especially lightweight, durable, and clear type of plastic. It has been used to make everything from food and drink containers, to car parts, to digital media and electronics. Phthalates have also become ubiquitous in a wide array of everyday products. Phthalates are a set of compounds that are added to plastic and vinyl to make materials more pliable. They are also a primary ingredient of many personal care products. In 1972, scientists discovered phthalates in human blood. In October 2007, the US Centers for Disease Control and Prevention (CDC) found that 93% of a random sample of 2,517 people had bisphenol-A in their urine.

Independent studies have linked these chemicals with endocrine disruption and are currently looking into the relationship between these chemicals and cancer, obesity, and hyperactivity. Other studies, funded primarily by the chemical industry, have found no causal link between bisphenol-A and phthalates and these deleterious effects. Some states have taken action to regulate these chemicals. California, Minnesota, Washington and Maryland have all proposed legislation to regulate phthalates. On Friday, April 18, 2008 Senator Charles E. Schumer (D-NY) reported that he was going to introduce legislation to ban the use of bisphenol-A in all children’s products and “food contact” consumer products—including water bottles and food containers. This proposed legislation is central to the main activity in this lesson: a hearing about whether or not the federal government should ban the use of bisphenol-A and phthalates in all consumer products.
Preparation:

Teachers should be familiar with the contemporary debates about bisphonol-A and phthalates by exploring the links in the Perspectives Section. This lesson also requires that student pairs have access to the internet. In addition to copies of documents listed below, students will use on-line sources to build and refine their arguments. This will ideally be done in a computer lab, or encourage students to bring laptops to class if they have them.

Activity 4.1: Ban Bisphenol-A?: Exploring a Contemporary Debate

In this activity, students will make arguments for or against the banning of bisphenol-A, or for alternative measures, by drawing on a variety of ideas and sources. Working in pairs, students will assume the identities of different perspectives including the American Chemistry Council, a European risk expert, health scientists, a historian, investigative journalists, an expecting parent, and others. Pairs will be given some primary documents to work with and links to additional resources to draw from. They will spend half the class period developing their argument and will also investigating counter-arguments by exploring other links. Teachers should encourage creative, but sound arguments.

In addition to the student pairs, three students will be selected to serve as the hearings’ Executive Committee. They will ultimately decide the outcome of the hearing. As others are building their cases during the first half of the course, the Executive Committee will familiarize themselves with as many perspectives as possible—though not to the same level of depth as the other groups. The Executive Committee will be responsible for running the hearings and being as fair and responsible to each party.

The second half of the class period will be devoted to the hearing itself. To begin, each pair will have 2 minutes to make their opening statement, including their stance on the proposed legislation and an explanation of this opinion. Then each group will have 2-3 minutes to explain their criticism of other group’s perspectives. Following the critique, there will be an open session moderated by the Executive Committee. The Committee should be prepared to ask questions or call on representatives to clarify certain points. At the end of the hearing, the Executive Committee will determine the course of future action.

Perspectives for Activity 4.1:

1. Anna Soto, M.D., Professor of Cell, Molecular and Developmental Biology, Tufts University
2. Steven Hentges, Ph.D., Executive Director, Polycarbonate/BPA Global Group, American Chemistry Council


3. Bill Durodie, Centre for Risk Management, King’s College London, also New College, University of Oxford


4. Susanne Rust, Meg Kissinger, and Cary Spivak, Investigative Journalists for the *Milwaukee Journal*


5. National Toxicology Program, Center for the Evaluation of Risks to Human Reproduction (CERHR)


6. Fred vom Saal, PhD, Developmental Biologist, University of Missouri.


7. Sarah Vogel, MPH, MEM, Ph.D., Historian, Columbia University


8. Hon. Andrew von Eschenbach, M.D., Commissioner, U.S. Food and Drug Administration


Background Material:
http://www.sciam.com/article.cfm?id=plastic-not-fantastic-with-bisphenol-a&print=true


Background Material:
http://www.rsc.org/chemistryworld/Issues/2008/April/BisphenolABabyBottleDebate.asp

11. Expecting Parent


Activity 4.2: Homework Assignment: Reflections on the Precautionary Principle

For this activity, students will need access to an internet connection. They will explore the Environmental Working Group’s Skin Deep: Cosmetic Safety Database site to check the toxicity of certain products that they use on a regular basis. The EWG website allows users to search specific products—everything from deodorant and toothpaste to nail polish and hair gel—to assess how hazardous they may be to users’ health.

To begin, have students write a list of personal care products that they used in the last 24 hours. The list should be as specific as possible, including brand names, particular scents or flavors, etc. Using the EWG website, students should search their products and write down each product’s “hazard score.” For the highest scoring product, have students click on the product and gather information on why the product received its score. The homework worksheet “Analyzing the Skin Deep: Cosmetic Safety Database” will help guide students through an evaluation of the website. It also
introduces them to the precautionary principle and asks students to write a short reflection essay about whether or not, on a smaller scale, they would apply the principle to their own decisions. That is, will they keep using the certain products once they know that they may be hazardous? Why or why not? The worksheet is provided below.
Homework Worksheet: Analyzing the Skin Deep: Cosmetic Safety Database*

*This website is produced by the Environmental Working Group

1. Write down a list of the personal care products that you used in the last 24 hours in the space provided below. Be as specific as possible, including brand names, particular scents or flavors, etc.

2. Click on the Environmental Working Group’s Skin Deep: Cosmetic Safety Database website: [http://www.cosmeticsdatabase.com/index.php?nothanks=1](http://www.cosmeticsdatabase.com/index.php?nothanks=1). In the search box, type in your products and write down the “hazard score” each product receives on your list above.

3. For your highest scoring product, click on the product to gather information on why the product received its score. To what types of hazards have ingredients from your product been linked?

4. On the right side of the screen, look at the “Ingredient Concerns” box. Why is there concern over the ingredients in this product? What is the breakdown in percentage of those concerns?

5. Scroll down to the Ingredients List and click on the one most highly ranked. Why is it ranked so high?

7. Go to the FAQ section of the Skin Deep site (there is a link to this section in the top right-hand corner). Explore some of EWG’s answers to questions about how these “hazard scores” are calculated. Do you trust this source? Why or why not?

8. In January 1998, a group of scientists, academics, and activists gathered at the Wingspread Conference Center in Racine, WI. Together, they drafted the Wingspread Precautionary Principle which states that:

   “Where an activity raises threats of harm to the environment or human health, precautionary measures should be taken even if some cause and effect relationships are not fully established scientifically.”

Write a short reflection essay about whether or not, on a smaller scale, you would apply this principle in your own decisions about the products you just learned about. That is, will you keep using products that may be potentially hazardous? Why or why not? Use issues and ideas we have discussed in the past few lessons to explain your answer.